
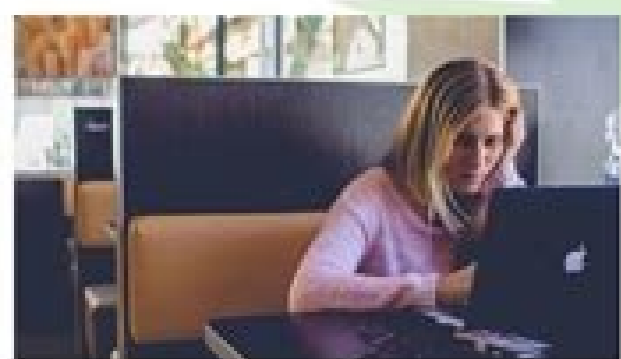


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3 Branches of Government Song Flocabulary
<https://www.flocabulary.com/3-branches-of-government/>

LEGISLATIVE:
 Yeah, it's the legislative branch. I'm the one that pass the laws. So if you want to make something happen, you've got to go through me.

EXECUTIVE:
 It's the executive branch. Who's calling the shots in the oval office?

JUDICIAL: Yeah, yeah, it's the courts. Judicial.
 It goes check, check to stay balanced,
 Not just one, we bring three talents
 The powers? Got to keep 'em separated,
 That's the motto, we made it.

3 branches of government, that's all:
 Legislative writes the laws,
 Executive executes laws — so official,
 The court system is Judicial

LEGISLATIVE:
 I'm the _____, I write laws,
 Without me, it'd be _____ of course.
 Say my name, I'm called _____
 Sitting on Capitol Hill, kid that's honest.
 Made up of the _____ and the _____

_____ and send it to the President's desk.
 We legislate, and we roll deep,
 535 members in my clique, see?

2 Four Strands of Science Learning

By looking much more deeply at how students learn science, recent research has produced new ways of thinking about what happens in science classrooms.

Books on science education have often drawn a fairly sharp distinction between scientific content and scientific processes. Content has been seen as the accumulated results of science—the observations, facts, and theories that students are expected to learn. Processes have been seen as the scientific skills that students are expected to master—skills like designing an experiment, making measurements, or reporting results.

Underlying the arguments in this book, however, is a new way of thinking about what it means to be proficient in science and a new framework for moving toward and achieving proficiency. This framework rests on a view of science as both a body of knowledge and an evidence-based, model-building enterprise that continually extends, refines, and revises knowledge. This framework moves beyond a focus on the dichotomy between content or knowledge and process skills, recognizing instead that, in science, content and process are inextricably linked.

This link between content and process is vital because scientific processes almost always take place when students are considering specific scientific content. When children use their ideas about the natural world to design investigations or argue about evidence, it strengthens their understanding of both the phenomena and the means used to investigate those phenomena. Moreover, separating content and process is inconsistent with what is now known about the way scientists actually do science.

Instead of drawing a distinction between content and process, we'll define and describe four learning "strands" that encompass the knowledge and reasoning skills that students eventually must acquire to be considered proficient in science.



Test taking flocabulary quiz answers. Test taking strategies flocabulary answers. Flocabulary test-taking vocabulary answers.

1051.2 grams⁵. Which of the following is greater?^a 500 grams of sugar occupies a volume of 0.315 liters. 68 kilograms⁹. Post-Assessment You should give the post-assessment to your students upon completion of all units of The Word Up Project to measure vocabulary growth from the pre-assessment. Finally, check to make sure your answer is reasonable. Feel free to email the results to . Heavy materials should have density values greater than that of water. When you return the test, you should celebrate how much students have learned since they took the pre-assessment. What is the mass of a 450 centimeter³ block of silicon if the density of silicon is 2.336 grams/centimeter³? Which liquid is on top? 26561 grams or 26.56 kilograms⁶. When both liquids are poured into a container, one liquid floats on top of the other. You can find the test, answer key and details on administering and interpreting it here. When you give the pre-assessment, you should not count the score toward students grades, as the purpose of the test is to determine how much students need to learn. Word Up Pre- and Post-Assessments Word Up Grade 2: Level Turquoise Word Up Grade 3: Level Red Word Up Grade 4: Level Orange Word Up Grade 5: Level Indigo Word Up Grade 6: Level Green Word Up Grade 7: Level Yellow Word Up Grade 8: Level Blue Word Up Pre- and Post-Assessment Answer Keys Word Up Grade 2: Level Turquoise Word Up Grade 3: Level Red Word Up Grade 4: Level Orange Word Up Grade 5: Level Indigo Word Up Grade 6: Level Green Word Up Grade 7: Level Yellow Word Up Grade 8: Level Blue Diagnostic Test The diagnostic test is intended to help you determine the appropriate Word Up Project level for your class. The density of a substance is 1.63 grams per milliliter. 8.33 pounds (2.2 kilograms = 1 pound, 1 liter = 0.264 gallons)¹⁰. However, if the majority of your class scores better than 80% on the pre-assessment, you may want to consider using the next level of The Word Up Project. 559 milliliter⁴. These liquids have different densities. If you choose to do classroom tracking, you can share the students averages with them, but it is not necessary to review the answers at this time, as students will learn the words over the course of The Word Up Project. What is the density of the sugar in grams per milliliter? If students do very well on the pre-assessment, it is important to stress that there are many other words that they will still learn, and that they will have the opportunity to become experts on the words they already know. What volume does 5 kilograms of copper occupy? 483.6 centimeters³ When you're asked to calculate density, make sure your final answer is given in units of mass (such as grams, ounces, pounds, kilograms) per volume (cubic centimeters, liters, gallons, milliliters). We'd also love to know how well the program worked for your students. (0.75 grams per milliliter)⁸. You can also use the results of this test as a final data point on a class- or student-tracking chart. You'll find the answers to each question at the bottom of the page. Make sure to tell them that you don't expect them to know all the words. So, if you have four significant digits for mass but only three significant digits for volume, your density should be reported using three significant figures. The number of significant figures will be the same as the number in your least precise value. The density of pure solid copper is 8.94 grams per milliliter. What is the mass of 0.25 liters of the substance in grams? The other thing to watch is the number of significant figures in your answer. 1.587 grams per milliliter². 7.8 grams per milliliter or 4.1 μg/μLb. 3 x 10⁻² kilograms/centimeters³ or 3 x 10⁻¹ milligrams/centimeter³ Two liquids, A and B, have densities 0.75 grams per milliliter and 1.14 grams per milliliter, respectively. Liquid A. a. You may be asked to give an answer in different units than you're given. Stephen Oliver / Getty Images This is a collection of 10 chemistry test questions with answers dealing with the density of matter. What is the mass of a 15 centimeter cube of iron if the density of iron is 7.87 grams/centimeter³? 1. It's a good idea to be familiar with how to perform unit conversions when working on these problems. One way to do this is to mentally compare your answer against the density of water (1 gram per cubic centimeter). If you tell your students that it does not count for a grade, explain that they should take the assessment seriously. At the end of the unit, you will be able to measure vocabulary growth by comparing results on the pre- and post-assessment. The pre-assessment measures your students vocabulary levels before they begin The Word Up Project, and is presented in the same format and level of difficulty as the post-assessment. 7.8 grams per milliliter b. How many kilograms of mercury would fill a 5-liter container if the density of mercury is 13.6 grams/centimeter³? 3 x 10⁻² kilograms/centimeter³7. Light substances would float on water, so their density should be less than that of water. 407.5 grams³. If you choose to set up a student- and class-tracking system, the pre-assessment will be the first data point to track. How much does 1 gallon of water weigh in pounds?Given: Density of water = 1 gram/centimeter³ How much space does 1 pound of butter occupy if the density of butter is 0.94 grams/centimeter³?

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